

COLLEGE OF FOOD, AGRICULTURAL, AND ENVIRONMENTAL SCIENCES

Understanding your water quality analysis report

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Water Quality workshop for fish farmers

Piketon, Ohio



Why test WQ?

- ☐ Most fish kills are caused by low dissolved oxygen.
- ☐ Waters from wells or springs are often devoid of dissolved oxygen.
- May also contain high levels of dissolved gasses, such as carbon dioxide and hydrogen sulfide, or metals, such as iron and manganese.
- ☐ It is important to know the source of water sample to interpret the results correctly.



When to test WQ?

□ Daily/Weekly/Bi-Weekly/Monthly/Annual

- ☐ For accurate results, water quality analyses must be performed within a certain time period after sampling.
- ☐ Holding times vary among parameters, typically ranging from 8 hours to 28 days.

How to collect samples?

| A clean plastic bottle with a screw cap thoroughly |
|---|
| rinsed with water from the source to be tested |
| Partially fill the bottle, cap it, and shake it a few times and discard away from sampling location |
| Completely fill the bottle with water & tightly seal |
| |
| Properly label the water sample |
| Collect at least two or three samples |
| ☐ If not possible - Use a composite sample |



How to collect samples? Do's & Don'ts

- ☐ Send samples to the laboratory as soon as possible
- ☐ Store in cool storage... preferably on ice
- ☐ Collect samples at depth of about ½ to 1 foot, in a location out several feet from the edge of the pond

- Avoid contamination from debris, mud, or algae in surface films
- Don't expose to the sun
- Avoid delays in shipping /delivery

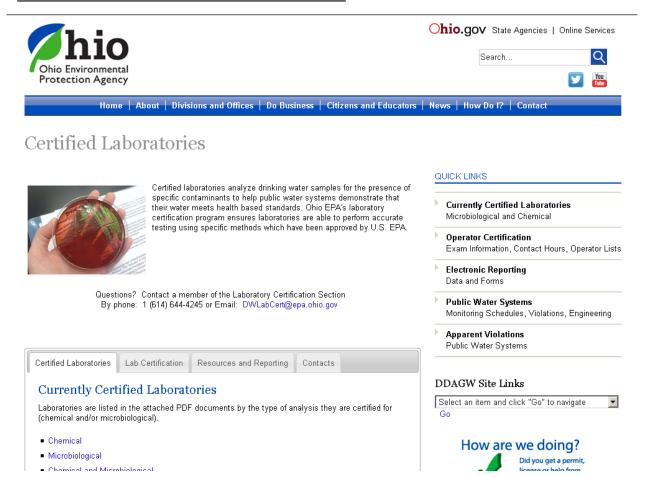


Sensitivity of parameters to time of shipment

- ☐ Parameters **likely to change** during the shipment:
 - □ pH, Electrical Conductivity, Fe (Iron) and Mn (Manganese), NH₃-N (Ammonia-Nitrogen), NO₂ (Nitrite) and NO₂-N (Nitrite-Nitrogen), PO₄ (Phosphate)
- ☐ Parameters **not likely** change during the shipment:
 - Alkalinity, Hardness,

Where to send samples for testing?

- Certified labs: http://www.epa.state.oh.us/ddagw/labcert.aspx
- OSU: www.southcenters.osu.edu/soil





OSU Resources

Piketon, OH
 OSU South Centers (Soil, Water, & BioEnergy Lab)
 Columbus, OH
 ?
 Wooster, OH
 STAR Lab

☐ OSU does not have commercial lab



Piketon: OSU South Centers

















WQ Report - Examples

| Parameter analyzed | Results | units |
|---|---------|----------------------------------|
| Calcium (Ca) | 76 | ppm |
| Magnesium (Mg) | 18 | ppm |
| Sodium (Na) | 979 | ppm |
| Potassium (K) | 71 | ppm |
| Boron (B) | 4.02 | ppm |
| Carbonate (CO ₃) | 0 | ppm |
| Bicarbonate (HCO ₃) | 367 | ppm |
| Sulfate (SO ₄ calculated from total S) | 601 | ppm |
| Chloride (CI-) | 1471 | ppm |
| Nitrate-N (NO ₃ -N) | 11.15 | ppm |
| Phosphorus (P) | 0.06 | ppm |
| pH | 7.33 | |
| Conductivity | 4680 | µmhos/cm |
| Hardness | 15 | grains CaCO ₃ /gallon |
| Hardness | 264 | ppm CaCO ₃ |
| Alkalinity | 301 | ppm CaCO ₁ |
| Total Dissolved Salts (TDS) | 3598 | ppm |
| SAR | 26.2 | |
| Charge Balance (cation/anion*100) | 81 | |

| Water Analysis Report: Fish |
|---|
| (Example) |
| pH7.24 |
| Electrical Conductivity 205 μSiemens/cm Alkalinity, Total 100.00 mg/l as CaCO ₃ |
| Hardness, Total 103.80 mg/l as CaCO ₃ |
| CO ₃ (Carbonate) 0.09 mg/l as CaCO ₃ HCO ₃ (Bicarbonate) 99.90 mg/l as CaCO ₃ |
| Fe (Iron)0.05 mg/l Mn (Manganese)0.01 mg/l |
| F (Fluoride)0.15 mg/l |
| Cl (Chloride)3.13 mg/l |
| SO ₄ (Sulfate) · · · · · · · · · · 3.21 mg/l NO ₃ (Nitrate) · · · · · · · · · · · 0.04 mg/l |
| NO ₃ -N (Nitrate-Nitrogen) 0.01 mg/l |
| NH ₃ -N (Ammonia-Nitrogen) 0.11 mg/l NO ₂ -N (Nitrite-Nitrogen) 0.00 mg/l |
| PO ₄ (Phosphate) 0.06 mg/l |

There is no standard list of water quality parameters that are measured, and units of measure will vary among laboratories.



Ranges of water quality parameters for fish

| Parameters | Desirable Range | Acceptable Range |
|---|---------------------------------------|--|
| рН | 6.5-9.5 | 5.5-10.0 |
| Electrical Conductivity | 100-2,000µSiemens/cm | 30-5,000µSiemens/cm |
| Alkanlinity (Total) | 50-150 mg/l as CaCO ₃ | 20-400 mg/l for ponds >10 mg/l for hatchery water as CaCO ₃ |
| Hardness (Total) | 50-150 mg/l as CaCO ₃ | > 10 mg/l as CaCO ₃ |
| CI (Chloride) | Above 60 mg/l (for catfish ponds) | 10 times the nitrite concentration (for catfish ponds) |
| NH ₃ -N (Ammonia- Nitrogen) | Total NH ₃ -N: 0-2 mg/l | Total NH ₃ -N: Less than 4 mg/l |
| | Un-ionized NH ₃ -N: 0 mg/l | Un-ionized NH ₃ -N: Less than 0.4 mg/l |

Ranges of water quality parameters for fish

| Parameters | Desirable Range | Acceptable Range |
|---|--|---|
| Fe (Iron) and Mn (Manganese) | For hatchery water: Ferrous iron: none Ferric iron: none | For hatchery water: No ferrous iron Ferric iron: Less than 0.1 mg/l for fry, less than 1.0 mg/l for most fish |
| | For pond water: Any level of ferric iron, no ferrous iron. | For pond water: Any level of ferric iron, presence of ferrous iron acceptable if limited to zone around water Inlet |
| | Manganese: 0-0.01 mg/l | Manganese: Up to 1 mg/l |
| NO ₂ (Nitrite) and NO ₂ -N (Nitrite-Nitrogen) | 0-1 mg/l NO ₂ | < 4 mg/l NO ₂ |



Ranges of water quality parameters for fish

| Parameters | Desirable Range | Acceptable Range |
|---|---|--|
| F (Fluoride) | Levels at or above 3 mg/l a losses of some fish species complex water conditions. | • |
| SO ₄ (Sulfate) | A wide range of sulfate con | ncentrations |
| CO ₃ (Carbonate) and HCO ₃ (Bicarbonate) | The relative amount of eac depends upon the pH of the | • |
| NO ₃ (Nitrate) and NO ₃ -N (Nitrate-Nitrogen) | Nitrate is relatively nontoxic not a health hazard except (above 90 mg/l NO ₃ -N) | |
| PO ₄ (Phosphate) | Phosphorus is an essential because it is often in limited phosphorus to water will stigrowth. This growth of algain the case of pristine stream ponds for fish culture. | d supply, adding imulate plant (algae) as can be undesirable, as |

Importance of the water quality for fish:

| A low pH (<4.5) - there is strong mineral acidity, which is harmful to fish & difficult/expensive to neutralize. |
|---|
| As pH increases, risk of ammonia toxicity to fish. |
| Some minimum salt content is desirable to help fish maintain their osmotic balance. |
| The amount of calcium hardness is important in pond fertilization because higher rates of phosphorus fertilizer are required at higher calcium hardness concentrations. |
| Chloride is a common component of most waters and is beneficial to fish in maintaining their osmotic balance. |



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