

Aquaponics: Water Quality

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Aquaculture Boot Camp 2

OSU South Center

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Important Water Quality Parameters

- ◆ Dissolved oxygen (DO)
- ◆ Nitrogen (ammonia, nitrite, nitrate)
- ◆ pH and alkalinity
- ◆ Carbon dioxide
- ◆ Settleable and suspended solids
- ◆ Water temperature

Source Water

- ◆ Well water
- ◆ Municipal water (must be dechlorinated)
- ◆ Rain water
- ◆ Surface water (not recommended)
- ◆ Obtain a water quality profile



Dissolved Oxygen

- ◆ Maintain DO levels at 5 mg/liter or higher
- ◆ Measure DO frequently in new system, but after procedures become standard it is seldom measured
- ◆ Tilapia will come to the surface for oxygen at 1 mg/liter DO
- ◆ DO is important for optimal fish and plant growth and for beneficial bacteria
- ◆ Aeration vs Oxygenation
- ◆ Water holds less oxygen at higher temperatures
- ◆ Measure DO with a meter









Ammonia

- ◆ Ammonia is excreted from fish gills
- ◆ The sum of the gaseous and ionic forms is called total ammonia-nitrogen (TAN)
- ◆ Ammonia is toxic to tilapia at 1mg/liter as a gas NH_3 (ammonia)
- ◆ Ammonia is not toxic as an ion NH_4 (ammonium)
- ◆ Equilibrium driven by pH and Temp
- ◆ At pH 7.0 or below most ammonia (>95%) exits as an ion
- ◆ Maintain TAN < 3 mg/liter

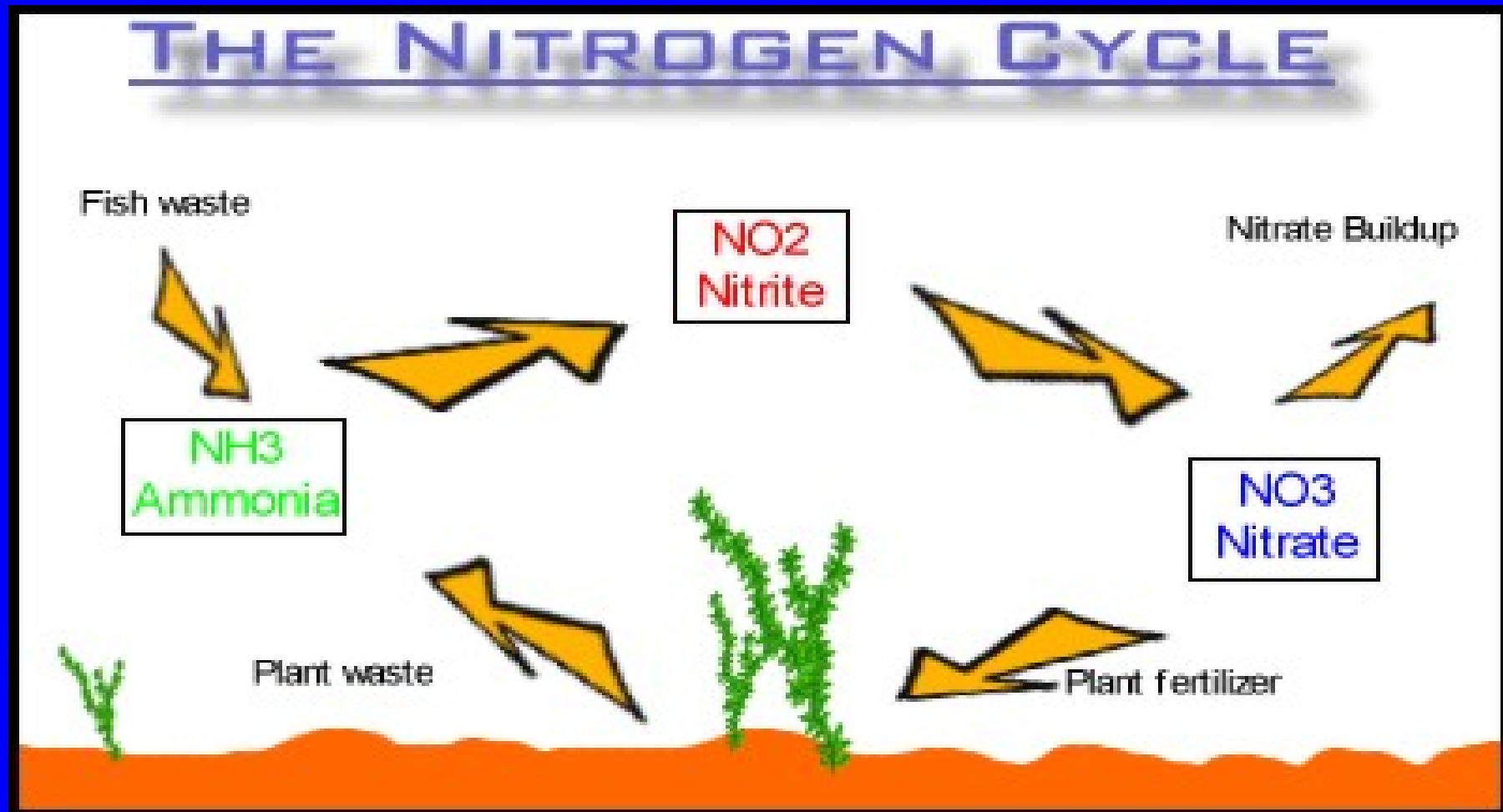
Ammonia Removal

- ◆ TAN is removed by nitrifying bacteria (*Nitrosomonas*) and plants
- ◆ Ammonium is oxidized to nitrite (NO_2)
- ◆ This process (nitrification) destroys alkalinity, produces acid (H) and lowers pH
- ◆ Nitrifying bacteria grow on surfaces (fixed film) or on suspended organic particles
- ◆ Nitrification is optimal at high DO and low levels of organic matter

Nitrite Removal

- ◆ Nitrite is removed by nitrifying bacteria (*Nitrobacter*)
- ◆ Nitrite is oxidized to nitrate (NO_3)
- ◆ Nitrate is relatively non-toxic
- ◆ This process (nitrification) also destroys alkalinity, produces acid (H) and lowers pH
- ◆ Nitrite is toxic at 5 mg/liter
- ◆ For tilapia maintain nitrite-nitrogen at 1 mg/liter

Nitrogen Cycle



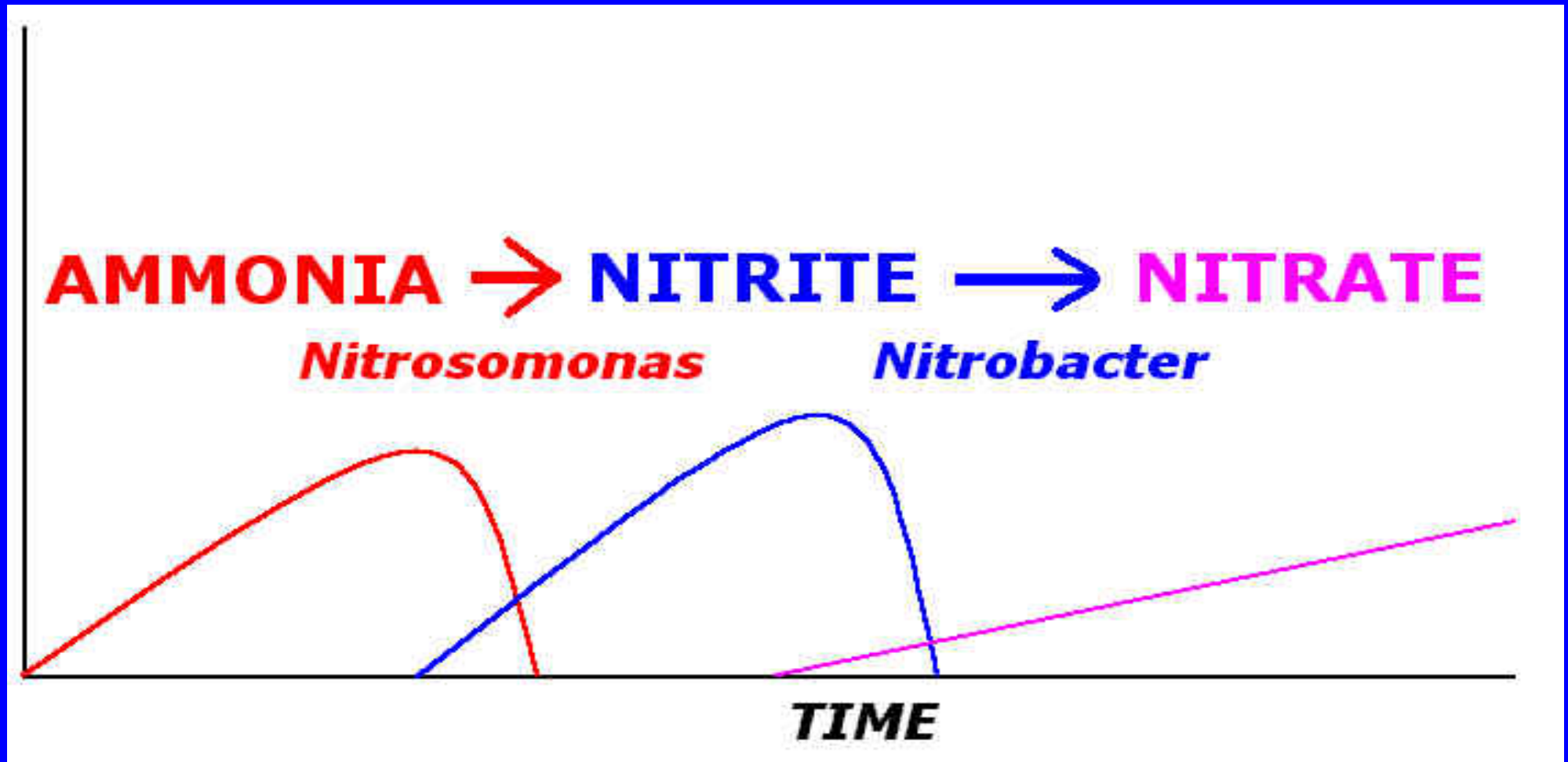




System Establishment

- ◆ Biofilters require 4-6 weeks for sufficient bacteria populations to develop
- ◆ Fish must be fed at a very low rate
- ◆ Measure ammonia and nitrite daily
- ◆ Ammonia levels rise and then decline followed by a rise and decline in nitrite levels
- ◆ Do not let TAN or $\text{NO}_2\text{-N}$ exceed 5 mg/liter
- ◆ Biofilter is established when TAN and nitrite levels decline and nitrate levels increase
- ◆ Reduce TAN and $\text{NO}_2\text{-N}$ monitoring
- ◆ Biofilter can be established with ammonia compounds before fish are added (SRAC 4502)

Acclimation Period

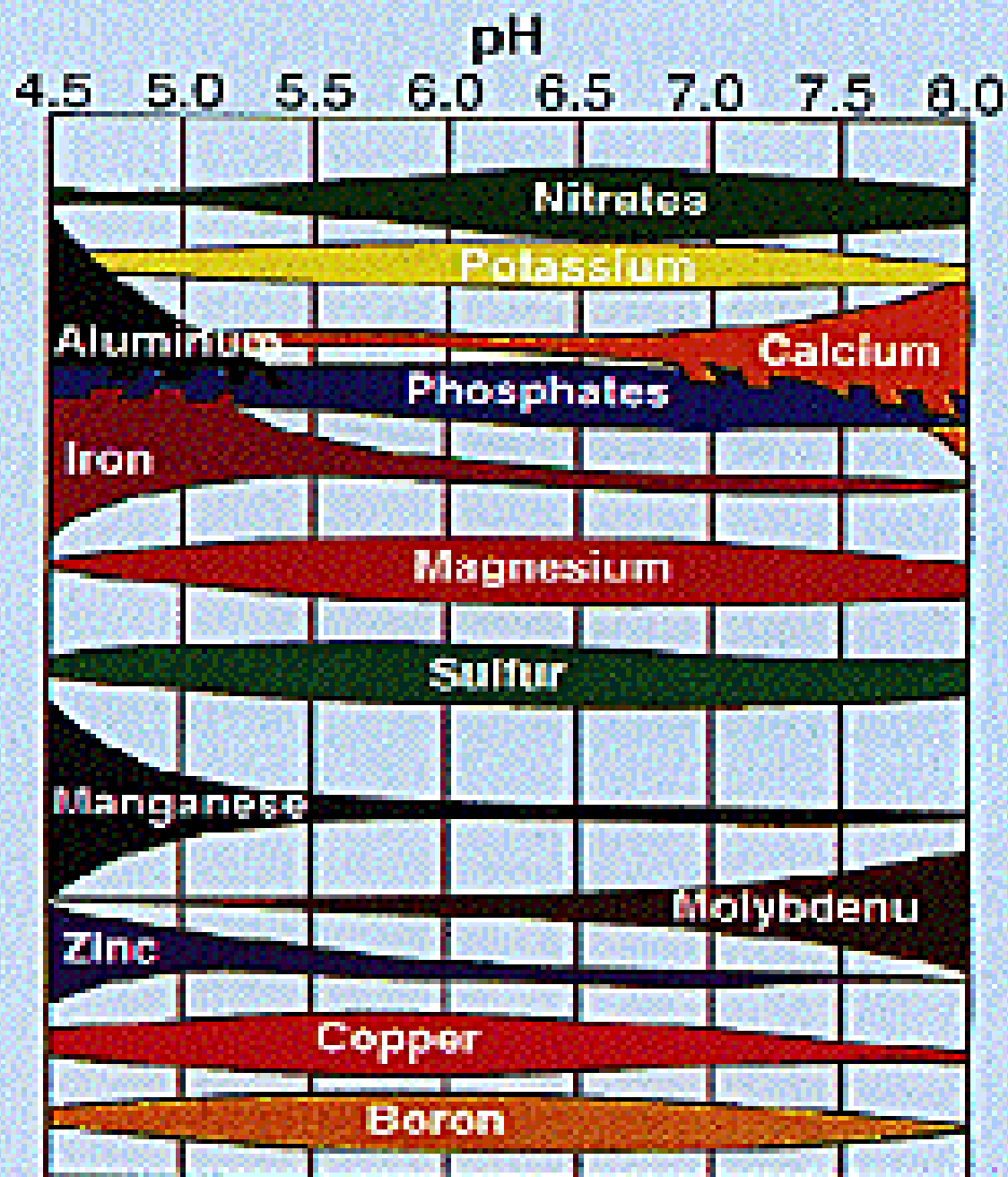


Denitrification

- ◆ Under anaerobic conditions (no oxygen) denitrifying bacteria convert nitrate (NO_3) to nitrogen gas (N_2)
- ◆ Denitrification produces alkalinity
- ◆ Decreasing nitrate levels is beneficial when raising fruiting plants such as tomatoes
- ◆ High nitrate levels promote vegetative growth and reduce fruit set
- ◆ High nitrate levels stimulate the growth of leafy green plants

pH

- ◆ Hydrogen ion activity (0-14)
- ◆ $0 < 7$ acidic, 7 -neutral, $7 < 14$ -basic
- ◆ Known as the master variable because it influences many water quality parameters (% of NH_3 vs. NH_4 and the solubility of plant nutrients)
- ◆ pH range for tilapia (5.0-10.0)



pH

- ◆ Plants prefer pH <6.5
- ◆ Nitrifying bacteria perform optimally at pH >7.5
- ◆ Compromise and maintain pH at 7.0
- ◆ Measure pH daily
- ◆ pH generally declines daily
- ◆ Add base to maintain pH 7.0
- ◆ Alternate adding calcium hydroxide with potassium hydroxide in base addition tank

Measuring pH



Low pH

- ◆ Sometimes operators neglect to measure pH for several days
- ◆ pH can quickly decrease to 4.5
- ◆ At pH 4.5 nitrification has ceased and TAN concentrations can exceed 30 mg/liter
- ◆ Add base very slowly over several days
- ◆ Adding a large amount of base at one time will shift the majority of TAN into the toxic gaseous form (NH_3) and kill all the fish

Alkalinity

- ◆ Alkalinity is a buffer that neutralizes acid
- ◆ Add base to increase alkalinity
- ◆ It's concentration is expressed as the equivalent concentration of calcium carbonate (CaCO_3)
- ◆ Maintain alkalinity >100 mg/liter as CaCO_3

Carbon Dioxide

- ◆ Carbon dioxide (CO₂) levels should not exceed 20 mg/liter
- ◆ At higher levels the fish become sluggish and cannot absorb enough oxygen through their gills
- ◆ In systems with diffused aeration CO₂ buildup is not a problem because it is vented off to the atmosphere
- ◆ CO₂ buildup can be a problem in systems using pure oxygen

Settleable Solids

- ◆ Settleable solids (feces, uneaten feed, biological growth) are large enough to settle to the tank bottom in a short time period
- ◆ Settleable solids are removed in the first stage of the solids removal component
- ◆ For passive removal, clarifiers and swirl separators are recommended
- ◆ Male tilapia fingerlings are required to scrape sides and concentrate solids at tank bottom
- ◆ A 20 minute retention time is required for clarifiers







Suspended Solids

- ◆ Fine solids that do not settle are removed in filter tanks
- ◆ Suspended solids adhere to orchard netting
- ◆ Filter tanks are cleaned periodically
- ◆ As these organic solids decompose essential inorganic nutrients are released to the culture water and taken up by plant roots
- ◆ This process is called mineralization
- ◆ Dissolved organic matter is also removed by the bacteria growing in the filter tanks

Water Temperature

- ◆ Fish species are temperature dependent
- ◆ Tilapia prefer 28-30°C for maximum growth
- ◆ Tilapia growth slows dramatically under 21°C and reproduction stops
- ◆ Tilapia die under 10°C
- ◆ Incidence of disease increases at lower temperatures as fish are stressed
- ◆ Vegetables prefer 21-24°C

Too Much Heat!



Feeding Rate Ratios Raft Systems

- ◆ Optimum feeding rate ratio is 60-100 g of fish feed/m² of plant growing area/day
- ◆ Use lower ratios for lettuce and staggered crops
- ◆ Use higher ratios for fruiting plants and batch culture
- ◆ These ratios generally supply 10 of the 13 nutrients required by plant crops

Nutrient Supplementation

- ◆ Add calcium hydroxide and potassium hydroxide to increase pH and supplement calcium and potassium
- ◆ Alternate additions
- ◆ Add equal amounts determined by trial and error
- ◆ Add 2 mg/liter of chelated iron (as iron) every 3 weeks or as needed
- ◆ Use chelated iron designated as DPTA

Nitrate Regulation

- ◆ Nitrate levels regulated by frequency of filter tank cleaning
- ◆ Frequent cleaning (twice a week) to increase nitrate levels – good for leafy greens
- ◆ Less frequent cleaning (once a week) to decrease nitrate levels – good for fruiting plants

Measurement of Total Nutrients Concentrations

- ◆ Use truncheon dip stick or EC meter
- ◆ These measure electrical conductivity (EC) and total dissolved solids (TDS)
- ◆ EC should range from 0.3 to 3.0
- ◆ TDS should range from 200 to 2,000 mg/liter
- ◆ EC and TDS are at the lower end of the range in aquaponic systems because nutrients are generated constantly



Water Quality Testing

- ◆ Test Kits
- ◆ Meters
- ◆ Spectrophotometers
- ◆ Certified labs



Contact Information



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